CO₂ Hydrogenation to Methyl Formate using CO₂ as the Sole Carbon Source

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The atmospheric concentration of CO_2 is still increasing, despite international efforts to curb emissions. The removal of CO_2 from the atmosphere is a high-cost endeavour, necessitating economic incentives for carbon capture. More importantly, the use of the near-unlimited C_1 resource of atmospheric CO_2 as a building block for process chemicals is essential to a sustainable future.¹ We have developed efficient catalysts for the hydrogenation of CO_2 to formic acid (FA) and methanol. Additionally, we have combined these C_1 building blocks into methyl formate (MF) in one system. Our catalysts are based on the encapsulation of noble metal complexes and subnanoparticles in the pores of zeolites, leveraging the acidity of the support and the principle of nanoconfinement to accelerate these multi-step reactions.



A Ru-based catalyst was developed for the hydrogenation of CO_2 to FA, and simultaneously facilitates fast esterification to MF (TOF > 100 h⁻¹), comparable to similar catalysts which operate at higher temperatures and pressures.^{2,3} A second series of catalysts, both from the literature and developed in our lab, are capable of CO_2 and FA hydrogenation to MeOH under the applied conditions. Combining two appropriate catalysts, MF in which both C atoms derive from CO_2 is formed. In the presence of a methoxide additive, the MF can be decarbonylated to give high CO concentrations (>20%) at a much lower temperature than traditional reverse water gas shift reactions.⁴ The structures of these catalysts will be discussed, and the reaction mechanisms will be explored to understand the key catalyst features which enable their reactivity. The present work allows the formation of three major C_1 synthons widely used in industry using CO_2 under relatively mild conditions. An outlook on coupling CO_2 hydrogenation with C-C bond-forming reactions will be described, as we progress towards the sustainable and selective synthesis of C_{2+} products from atmospheric carbon.

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